

Annual Meeting on Reaction Engineering 2023 15 - 17 May 2023 DECHEMA-Haus, Frankfurt am Main



Digitalization in Catalysis and Reaction Engineering: More than just a Buzzword?

Olaf Deutschmann, KIT, Germany

Stephan A. Schunk, hte GmbH, BASF SE, Germany



Digitalization in science and engineering Anything else than new



Konrad Zuse and his first computer, the Z3, from 1941



Source: Deutsches Museum https://www.ingenieur.de/technik/produkte/konrad-zuses-z3-computer-welt-80 accessed 3.5.2023

First computer codes modeling reactive flows, 1979

CHEMKIN: A General-Purpose, Problem-Independent, Transportable, **Fortran Chemical Kinetics Code Package**

THERMO 0 1H 1 G 0300.00 5000.00 1000.00 OH 0.02882730E+02 0.10139743E-02-0.02276877E-05 0.02174683E-09-0.05126305E-14 0.03886888E+05 0.05595712E+02 0.03637266E+02 0.01850910E-02-0.16761646E-05 0.02387202E-07-0.08431442E-11 0.03606781E+05 0.13588605E+01

R. J. Kee, J. A. Miller, T. H. Jefferson

Prepared by Sandia Laboratories, Albuquerque, New Mexico 87115 and Livermore, California 94550 for the United States Department of Energy under Contract DE-AC04-76DP00789.

Printed March 1980

east LENIW

east KK, the total number of species

SUBROUTINE CKGML (T. ICKWRK, RCKWRK, GML) Returns the standard state Gibbs free energies in molar units; see Eq. (24).

dard state gibbs free energies in molar

cgs units - kelvins cgs units - kelvins Data type - real scalar sy of integer workspace Data type - integer arri Dimension ICKWRK(*) at

of real work space. Data type - real array Dimension RCKWRK(=) at

the species. cgs units - ergs/mole Data type - real array

Temperature.

INPUT

Courtesy of R.J. Kee

SPECIES H2 H O2 O OH HO2 H2O2 H2O N N2 NO END REACTIONS H2 + O2 = 2OH 0.170E + 14 OH + H2 = H2O + H0.117E + 10 O + OH = O2 + H0.400E + 15 O + H2 = OH + H0.506E + 05 2.67 H + O2 + M = HO2 + M0.261E 1.10 OH+

ELEMENTS HONEND

	0.30 IE 1 10	-0.72	0	DINOIN-LEWIS
H2O/18.6/ H2/2.86/ N2/1.26/				
OH + HO2 = H2O + O2	0.750E + 13	0.00	0	! D-L
H + HO2 = 2OH	0.140E + 15	0.00	1073	! D-L
O + HO2 = O2 + OH	0.140E + 14	0.00	1073	! D-L
20H = 0 + H2O	0.600E + 09	1.30	0	! COHEN-WEST
H + H + M = H2 + M	0.100E + 19	-1.00	0	! D-L
H2O/0.0/ H2/0.0/				

0.00

1.30

0 72

-0.50

47780

3626

! D-L&W

6290 ! KLEMM ET AL., 1986

DIVON LEW

0 ! JAM 1986



Olaf Deutschmann and Stephan A. Schunk Digitalization in Catalysis and Reaction Engineering: More than just a Buzzword?

Digitalization in Catalysis and Reaction Engineering Numerical simulation of the behavior of chemical reactors



Hydrogen from methane by catalytic partial oxidation



 $CH_4 + \frac{1}{2}O_2 \rightarrow 2H_2 + CO$ $CH_4 + 2O_2 \rightarrow 2H_2O + CO_2$



R. Schwiedernoch, S. Tischer, C. Correa, O. Deutschmann, Chem. Eng. Sci. 58 (2003) 633.

Digitalization in Catalysis and Reaction Engineering Numerical simulation of the behavior of chemical reactors



Hydrogen from methane by catalytic partial oxidation



Surface coverage



R. Schwiedernoch, S. Tischer, C. Correa, O. Deutschmann, Chem. Eng. Sci. 58 (2003) 633.



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Digitalization in Catalysis and Reaction Engineering Numerical simulation of the behavior of chemical reactors



Hydrogen from methane by catalytic partial oxidation





R. Schwiedernoch, S. Tischer, C. Correa, O. Deutschmann, Chem. Eng. Sci. 58 (2003) 633.

Laminar – turbulence transition behind a monolithic catalyst



M. Hettel, E. Daymo, T. Schmidt, O. Deutschmann. Chem. Eng. & Processing: Process Intensification 147 (2020) 107728.

Reaction rate in fixed bed reactor using micro kinetics



E. A. Daymo, M. Hettel, O. Deutschmann, G. D. Wehinger. Chem. Eng. Sci. 250 (2022) 117408.

Digitalization in Catalysis and Reaction Engineering National research data infrastructure







D. Linke, C. Wulf, M. Beller, T. Boenisch, O. Deutschmann, S. Hanf, N. Kockmann, R. Kraehnert, M. Oezaslan, S. Palkovits, S. Schimmler, S.A. Schunk, K. Wagemann. ChemCatChem 13 (2021) 3223.

 \rightarrow A. FedorovTue 10:00

Annual Meeting on Reaction Engineering 2023

Olaf Deutschmann and Stephan A. Schunk Digitalization in Catalysis and Reaction Engineering: More than just a Buzzword?

Digitalization in Catalysis and Reaction Engineering Outline



- Introduction –long history of digitalization in science and engineering (OD)
- Automated workflows on the lab scale models for heterogeneous catalysis (OD)
- Feedback loops –from lab to pilot plant (SAS)
- Application of digital tools for scale-up on industrial scale (OD)
- Prediction of materials properties on atom scale using AI for catalyst discovery (SAS)
- Democratization of tools and services: RDM @ NFDI4Cat (SAS)

Variety of experimental methods for a better understanding of chemical kinetics and interaction with mass and heat transport



End-of-pipe



Fixed bed reactor

Laser diagnostics



Catalytic wall reactor

Capillary techniques



Monolithic reactors

Specific flow reactors



High-temperature pressure reactor



Stagnation flow reactor

Electrochemical characterization



Software package DETCHEM for numerical simulation of a wide variety of chemical reactors





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Digitalization in Catalysis and Reaction Engineering: More than just a Buzzword?

Modeling of heterogeneous chemical reactions: **Development of reaction mechanisms with associated kinetic data**





Bundesministeriun

für Wirtschaft und Technologie

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Digitalization in Catalysis and Reaction Engineering: More than just a Buzzword?

Automated workflows in catalysis and reaction engineering: Speed-up of model development by CaRMeN





....

973K

873K

973K

Experimental data including all metadata for different reactors, conditions, sources for given chemical system

Characterization

Simulation meta data of experiments data of physical & chemical models CaRMe ▲ 巫 Annular EOP Rh CH4 CO O2 Detailed Comparison Parity Plot Rh/Al2O3 Steam Deutschmann 20010213 Reforming CH4/H20=1.0/2.0 Annular EOP Rh/Al2O3 Steam Reforming CH4/H2O=1.0/2.0 Annular EOP Rh/Al2O3 Steam Reforming CH4/H2O=1.0/2.0 Rh CH4 O2 Spaci-Pro Rh/Al2O3 Hickman 1993 Driver: detchem channel Driver: detchem_plug CH4 CPOX 792K Rh CH4 CO2 O2 Spaci-Pro Rh/Al2O3 Karakaya 2016 CH4 CPOX 500K Rh CH4 H2O Spaci-Pro Rh/Al2O3 Kechagiopoulos 0 2017 CH4 CPOX 797K Spaci-Pro Rh/Al2O3 Rh CH4 H2O O2 Schaedel 2009 CH4 CPOX 798K Stag Rh CH4 steam 1008K Stag Rh CH4 steam 🗏 H2O 📕 H2 📕 CO 🗏 CO2 H20 H2 CO CO2 Spaci-Pro Rh/Al2O3 CH4 CPOX 792K Stag Rh CH4 CPOX Stag Rh CH4 CPOX 973K C-O=1 total 1023K Driver: detchem_channel Driver: detchem_stagnation Stag Rh CH4 CPOX 0 Stag Rh CH4 CPOX Model vs. Experiment 973K C-O=1 total Stag Rh CH4 CPOX 973K rich for total 0 oxidation Automated comparison of Stag Rh CH4 CPOX experimental and/or stoichiometric for total oxidation ^{co2} numerically simulated[∞] data¹² • ^{o2} • ^{H20} Rh S/C=1 GHSV=70000 1/h Rh S/C=1, GHSV=70000 1/h Rh S/C=1. GHSV=90000 1/h Driver: detchem_channel Driver: detchem_channel Rh S/C=1 GHSV=90000 1/h

Combines chemical and physical models

Models

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and can be used for easy comparisons of different

- reaction mechanisms.
- transport models,
- computer codes ...

H. Gossler, L. Maier, S. Angeli, S. Tischer, O. Deutschmann. PhysChemChemPhys 20 (2018) 10857; Catalysts 9 (2019) 227; www.detchem.de/software/carmen

Olaf Deutschmann and Stephan A. Schunk Digitalization in Catalysis and Reaction Engineering: More than just a Buzzword?

Automated workflows in catalysis and reaction engineering: CaRMeN as data basis for model parametrization and optimization





Olaf Deutschmann and Stephan A. Schunk

Digitalization in Catalysis and Reaction Engineering: More than just a Buzzword?

Digital twin of reactor and direct transfer of performance and meta data: Software tool Adacta

Karlsruhe Institute of Technology

- Focus on traceability of data
- Captures exact details to an experimental setup
- Setups composed of devices and samples
- **Time line** as independent variable to visualize relationships
- Data exactly associated with recording device
- Catalyst/materials history monitored including efficient search tools





Data archive and extraction of metadata for setting up input files for models and simulation

H. Gossler, J. Riedel, E. Daymo, R. Chacko, S. Angeli, , O. Deutschmann. Chemie Ingenieur Technik 94 (2022) 1798; www.omegadot.software/adacta

Digitalization in Catalysis and Reaction Engineering: More than just a Buzzword?

Digitalization in catalysis and reaction engineering: Value-added coupling of digitalization tools





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Feedback Loops: Cause and Effect made simple?



"...Simple causal reasoning about a feedback system is difficult because the first system influences the second and second system influences the first, leading to a circular argument. This makes reasoning based upon cause and effect tricky, and it is necessary to analyze the system as a whole. ..."

— Karl Johan Åström and Richard M.Murray, Feedback Systems: An Introduction for Scientists and Engineers

Feedback Loops in Action





From Naphta to Reformate





- Detection of very small differences in product selectivity (C₅+)
- Variation of feed & product properties
- Accelerated decay conditions for fixed-bed catalysts
- Collection of sufficient data for fast deactivating CCR catalysts



Why Testing under "Iso"-Conditions?



- Comparability on a relative scale
- Comparability on an absolute scale
- Retrieval of kinetic data-sets that link into scale-up criteria

Complexity of the Analysis





hte Software Solutions Combining workflow key steps with one tool



GENERATING DATA





- Recipe editor
- Process control
- Industry standard process visualization
- Intuitive test rig operation
- Fast and efficient failure diagnosis



- Access data from your desktop
- All data in one database
- Evaluate and • create reports
- Integrate your **R&D** environment



ONLINE GC ANALYSIS

OFFLINE ANALYSIS

GENERATING KNOWLEDGE





& CALCULATIONS

AUTOMATED DATA MERGE

REPORTING & EVALUATIO

Iso-Operation Integrated Cyber-Physical Interaction



Performance Parameters



Global Peak Identification

- myhte allows interactive self-adaptation to prechosen values for distinct variables
- This is an excellent example for a self-adaptive cyber-physical interaction

hte-company.com

Feedback Loops in Action





Transalkylation of Aromatics





- Conversion of C7 and C9 to maximize xylenes (p-xylene)
- Scopes
 - Identification of best performing catalyst
 - · Deactivation at higher severities
 - Quality control

Multicomponent Mixtures Overlay of wavelengths: chemometrics







IR spectra

- GC chromatograms
 - Selectivities
 - Conversions
 - RON, MON

Transalkylation Prediction of C9+ conversion





Feedback Loop Achieving Constant Conversion





Fast Feedback Loop based on Chemometrics the Power of FTIR Spectroscopy





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- Application of digital tools for scale-up on industrial scale (OD)
 - High-temperature electrolysis
 - Carbon-free chemical energy carriers
 - CO₂ emission reduction in steel production
- Prediction of materials properties on atom scale using AI for catalyst discovery (SAS)
- Democratization of tools and services: RDM @ NFDI4Cat (SAS)

Hydrogen from high-temperature electrolysis: Process intensification trough coupling of thermal and chemical processes





Solid-Oxide Electrolysis Cell (SOEC) operated >700°C

Power-to-Methane: Thermal integration of SOEC stack & methanation reactor for



What is the best SOEC configuration for thermal integration?

- Stack simulation based on physical and chemical characterization of the SOE Cell
- Optimization of design parameters and operating conditions

L. Wehrle, D. Schmider, J. Dailly, A. Banerjee, O. Deutschmann. Appl. Energy 317 (2022) 119143.

Hydrogen from high-temperature electrolysis: Development chain from button cell to stack to system





L. Wehrle, D. Schmider, J. Dailly, A. Banerjee, O. Deutschmann. Appl. Energy 317 (2022) 119143.

→ Poster 4.02 Furst et al.

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Hydrogen from high-temperature electrolysis: Digitalization tool CaRMeN accelerates parametrization



characterization EIS Verbindung Massedurchfluss Strom/Snannu regler für Wasser Eingangsgase Anschluss Eingangsgase Prüfstand 7ellaehäuse

Electrochemical



Electrochemical model parameters

Parameter	Value						
Cell type	ESC	CSC					
Cathode material	Ni-CGO	Ni-YSZ					
Exchange current density	5.9×10 ³ ×	0.68×					
/ (A cm ⁻¹ or A cm ⁻²)	e ^{-111900/RT} ×p _{H2} ^{0.01} ×p _{H2O} ^{0.08}	e ^{-105000/RT} ×p _{H2} - ^{0.1} ×p _{H2O} ^{0.33}					
Anodic CT-coefficient	0.75	0.6					
Cathodic CT-coefficient	0.25	0.4					
Anode material	LSCF	LSC					
Exchange current density	8.5×10 ² ×e ^{-138000/RT} ×p ₀₂ ^{0.11}	7.6×10 ⁵ ×e ^{-150000/RT} ×p _{O2} ^{0.22}					
Anodic CT-coefficient	0.5	0.65					
Cathodic CT-coefficient	0.5	0.35					
	OEB: Owner Evolution	Position					
	OER: Oxygen Evolution	Reaction					
	$\begin{array}{c} \textbf{OER: Oxygen Evolution} \\ \hline O_0^{\infty} \leftrightarrow V_0^{\text{or}} + 2e^- + 0, \\ \hline \text{Air} \end{array}$	Reaction 5 O ₂ (g)					
CGO	OER: Oxygen Evolution $O_0^X \leftrightarrow V_0^{-} + 2e^- + 0$. Air Au grid	Reaction $5 O_2(g)$					
CGO Ceramic sealing	OER: Oxygen Evolution $O_0^x \leftrightarrow V_0^{**} + 2e^- + 0.$ Air Air Au grid LSC YSZ NU-YSZ (II) NU-YSZ (II)	Reaction 5 O ₂ (g)					
Go Ceranic	OER: Oxygen Evolution $O_0^{r} \leftrightarrow V_0^{r*} + 2e^- + 0.$ Air Air Au grid LSC NL-YS2 (d) H ₂ H ₂ H ₂	Reaction 5 O ₂ (g) Cell housing Ni mesh					
Go	OER: Oxygen Evolution $O_0^{r} \leftrightarrow V_0^{rr} + 2e^- + 0.$ Air Air Augrid LSC Vices (II) Ni-YSZ (II) H ₂ H	Reaction 5 O ₂ (g) Cell housing Himesh H ₂ + 00					

L. Wehrle, D. Schmider, J. Dailly, A. Banerjee, O. Deutschmann. Appl. Energy 317 (2022) 119143.

Next step: Automatic parameter adaptation → Poster 1.01 Jägerfeld et al.

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Digitalization in Catalysis and Reaction Engineering: More than just a Buzzword?

Hydrogen from high-temperature electrolysis: Digitalization and multi-scale modeling supports scale-up



Benchmarking SOEC-stack designs for industrial Power-to-Methane systems



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Digitalization in Catalysis and Reaction Engineering: More than just a Buzzword?

Reactive metals as carbon-free energy carriers for a circular energy economy: Iron cycle



 Metals: Promising solution for efficient storage of renewable energy in amounts of 100 TWh due to high volumetric energy density, easy handling, use of existing infrastructures



J.M. Bergthorson. Prog. Energy Combust. Sci. 68 (2018) 169.
P. Debiagi, J. Janicka, C. Hasse et al. Renewable Sustainable Energy Rev. 165 (2022) 112579.
C. Kuhn, A. Düll, P. Rohlfs, S. Tischer, M. Börnhorst, O. Deutschmann. Appl. Energy and Combust. Sci. 12 (2022) 100096.

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M. Fedoryk, B. Stelzner, S. Harth, S., D. Trimis. Appl. Energy and Combust. Sci. 13 (2023)

Annual Meeting on Reaction Engineering 2023

Iron as carbon-free energy carrier: Cycling demands a detailed understanding of kinetics and morphology



- TGA analysis: Cycling by oxidation with air and reduction with H₂ works
- Significant morphological and reaction rate variations



→ Poster 9.08 Kuhn, Knapp et al.

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Digitalization in Catalysis and Reaction Engineering: More than just a Buzzword?

Iron as carbon-free energy carrier: Monitoring and archiving all experimental details by digital twin using Adacta





Configuration of experimental setup

H. Gossler, J. Riedel, E. Daymo, R. Chacko, S. Angeli, , O. Deutschmann. Chemie Ingenieur Technik 94 (2022) 1798. www.omegadot.software/adacta

Olaf Deutschmann and Stephan A. Schunk Digitalization in Catalysis and Reaction Engineering: More than just a Buzzword?

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All information on setup, devices, samples, operating conditions, performance data et al. at any time accessible from web

H. Gossler, J. Riedel, E. Daymo, R. Chacko, S. Angeli, , O. Deutschmann. Chemie Ingenieur Technik 94 (2022) 1798. www.omegadot.software/adacta

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Reduction of CO₂ emission from flue gases of steel industry by dry reforming of methane and re-use in the process

- Steel industry is responsible for 7–9% of the global direct CO₂ emissions
- 70% of steel production by coke-dependent blast furnace basic oxygen furnace route
- Steel plant off-gases contain value components



W. Uribe-Soto, J.-F. Portha et al., Ren. Sust. Energy Rev. 74 (2017) 809.

CH₄ (vol%)	→ 22.0	0.0
$C_x H_y$ (vol %)	2.0	0.0
CO ₂ (vol %)	1.2	→ 21.6
H₂O (vol%)	4.0	4.0
H₂ (vol%)	60.7	3.7
CO (vol%)	4.1	23.5
N ₂ (vol %)	5.8	46.6
$Ar + O_2$ (vol%)	0.2	0.6
$Ar + O_2$ (vol%)	0.2	0.6

COG

Feed name

S. Angeli, S. Gossler, S. Lichtenberg, G. Kass, A. Agrawal, M. Valerius, K. P. Kinzel, O. Deutschmann. Angew. Chemie Intl. Ed. 60 (2021) 11852.



 $Fe_2O_3 \rightarrow Fe$

BFG

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Digitalization in Catalysis and Reaction Engineering: More than just a Buzzword?

Reduction of CO₂ emission from flue gases of steel industry by dry reforming of methane and re-use in the process

- Steel industry is responsible for 7–9% of the global direct CO₂ emissions
- 70% of steel production by coke-dependent blast furnace basic oxygen furnace route
- Use of high-temperature Coke Oven Gas (COG) and Blast Furnace Gas (BFG)
- "Dirty" components of COG and BFG would poison the catalyst
- Reforming of COG/BFG mixtures in the regenerative heat exchanger (cowper) at 800°C-1400°C
- Syngas use for iron ore reduction in blast furnace



P. Blanck, G. Kass, B. Kanz, K.P. Kinzel, O. Deutschmann. Energy Adv., subm .

S. Angeli, S. Gossler, S. Lichtenberg, G. Kass, A. Agrawal, M. Valerius, K. P. Kinzel, O. Deutschmann. Angew. Chemie Intl. Ed. 60 (2021) 11852.

Olaf Deutschmann and Stephan A. Schunk

Digitalization in Catalysis and Reaction Engineering: More than just a Buzzword?

$CH_4 + CO_2 \rightarrow 2H_2 + 2CO$

Feed name	COG	BFG
CH₄ (vol%)	22.0	0.0
C _x H _y (vol %)	2.0	0.0
CO ₂ (vol %)	1.2	→ 21.6
H ₂ O (vol%)	4.0	4.0
H ₂ (vol%)	60.7	3.7
CO (vol%)	4.1	23.5
N ₂ (vol %)	5.8	46.6
$Ar + O_2$ (vol%)	0.2	0.6



 $Fe_2O_3 \rightarrow Fe$

Reduction of CO₂ emission from flue gases of steel industry **Digitalization used for understanding and process development**

- Understanding reaction kinetics in the gas-phase and model development for scale-up
- Four models available in literature but none of them ever tested for these conditions

COG/REG=0.5 3bar 100Nm

- PolyMech [1]: 558 reactions, 83 species
- Golovichev [2]: 690 reactions, 130 species

cle34 15.03.22

- Konnov [3]: 1231 reactions, 129 species
 - AramcoMech [4]: 3037 reactions, 581 species

0.6

0.5

(siseq

PolvMech

Konnov Golovitchev



[1] S. Porras, C. Schulz, U. Maas et al., Combust. Flame 212 (2020) 107. [2] V. I. Golovitchev, F. Tao, J. Chomiak, SAE Tech. Pap. 1999-01-3552. [3] A. A. Konnov, Combust. Flame 156 (2009) 2093. [4] C.-W. Zhou, H.J. Curran et al., Combust. Flame 197 (2018) 423.

Karlsruhe Institute of Technology

COG/BFG=1

P=5.5 bar



S. Angeli, S. Gossler, S. Lichtenberg, G. Kass, A. Agrawal, M. Valerius, K. P. Kinzel, O. Deutschmann. Angew. Chemie Intl. Ed. 60 (2021) 11852.

1300

1400

Reduction of CO₂ emission from flue gases of steel industry Digitalization tools support scale-up and operation strategies

- Maximum reduction of 78% CO₂ can be achieved
- Potential of reducing 12 % of steel plant CO₂ emissions
- World-wide implementation \rightarrow 0.5% less global CO₂ emissions
- DETCHEM and CaRMeN now used for scale-up by industry



S. Angeli, S. Gossler, S. Lichtenberg, G. Kass, A. Agrawal, M. Valerius, K. P. Kinzel, O. Deutschmann. Angew. Chemie Intl. Ed. 60 (2021) 11852.

 \rightarrow Poster 9.06 Blanck et al.



<text>

P. Blanck, G. Kass, B. Kanz, K.P. Kinzel, O. Deutschmann. Energy Adv., subm.

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iterative process Catalyst synthesis (as bulk Model development (e.g., DFT) and material) and characterization candidate library generation Synthesis of supported Quantum Machine Learning algorithms to accelerate Catalyst catalysts of the most promising **Mechanics** QM simulations and analyze huge number of materials (including ALD) **Synthesis** structures and compositions Machine Learning

Case Study

Mixed Metal Oxides as Potential Candidates for CO₂ to MeOH

Scope: Development of a predictive tool for catalyst design by correlation of catalyst properties and performance Identification of a competitive catalyst technology for CO₂ to MeOH

Catalyst

Testing

Machine learning to find descriptor to correlate the structure properties with catalytic activity

ln_2O_3 CO3+3H

- Catalyst testing in high throughput plant: for fast screening and stability tests
- Kinetic studies in the profile reactor





Identify Opportunity by Finding Patterns



Theoretical volcano for the production of methane from syngas, CO, and $\rm H_2$





https://github.com/sandipde/Interactive-Sketchmap-Visualizer

hte-company.com

Identify Opportunity by Finding Gaps



G. Arthur Montreal, CC BY-SA 4.0 via Wikimedia Commons https://commons.wikimedia.org/wiki/File:First_Version_of_the_Periodic_Table.gif



http://interactive.sketchmap.org/

Comparing molecules and solids across structural and alchemical space, Sandip De et al. *Phys. Chem. Chem. Phys.*, 2016,18, 13754-13769

CO₂ to Methanol Cost Performance Tradeoff





Catalyst Performance

Probe into In₂O₃+ dopants





Utilize existing Databases for High Throughput Screening – Compute when necessary!





Calculate key descriptors like vacancy (Ov) formation energies and adsorption energies.



- Not everything is there in databases!
- Compute what we need quickly with high-throughput computation





Machine learning at Atomic Scale: Complexities of Interatomic Interactions





- ML model is 10³-10⁶x faster than reference QM
- Improving at each iteration and converges ~ 5-10 iterations
- ML training takes ~10-24 hours



Crystalline Solids predicted by Quantum Chemistry Simulations enabling Understanding of Structure-Performance Relationships



Candidate Elements for Mixed Metal Oxides in the Light of the Target Reaction

					-					_							
IA I																	18 VIIIA
1 Hydrogen	2 IIA											13 IIIA	14 IVA	15 VA	16 VIA	17 VIIA	² He Helium 4.002602
³ Li Lithium	Beryllium											5 Boron	6 Carbon	7 N Nitrogen	8 Oxygen	⁹ F	Ne
¹¹ Na	12 Magnesium	3 IIIB	4 IVB	5 VB	6 VIB	7 VIIB	8 VIIIB	9 VIIIB	10 VIIIB	11 IB	12 IIB	13 Aluminium	¹⁴ Si Silicon	15 P Phosphorus	¹⁶ S Sulfur	17 Cl Chlorine	18 Argon
¹⁹ K	°Ca	Sc	²² Ti	²³ V	²⁴ Cr	²⁵ Mn	Fe	²⁷ Co	²⁸ Ni	°°Cu	[®] Zn	Ga	³² Ge	³³ As	°Se	³⁵Br	³⁶ Kr
Potassium 39.0983	Calcium 40.078	Scandium 44.955908	Titanium 47867	Vanadium 50.9415	Chromium 51,9961	Manganese 54,938044	Iron 55,845	Cobalt 58,933794	Nickel 58,6934	Copper 63.546	Zinc 65.38	Gallium 69.723	Germanium 72,630	Arsenic 74.921595	Selenium 78.971	Bromine 79.904	Krypton 83,798
³⁷ Rb	^³ Sr	³⁹ Y	°r	^⁴ Nb	Mo	⁴³ Tc	^₄ Ru	^⁵ Rh	* Pd	Aq	°℃d	[®] In	ຶSn	ຶSb	⁵²Te	53	⁵́Xe
Rubidium	Strontium 8762	Yttrium 88.90584	Zirconium	Niobium 92.90637	Molybdenum	Technetium	Ruthenium	Rhodium	Palladium	Silver	Cadmium	Indium 114.818	Tin 118.710	Antimony 121,260	Tellurium	lodine 126.90447	Xenon 131,293
⁵⁵ Cs	Ba	57 - 71 Lanthanoids	™Hf	⁻³Ta	⁷⁴ W	° [™] Re	⁷⁶ Os	" Ir	⁷⁸ Pt	⁷⁹ Au	[®] Hg	⁸¹ TI	°₽b	³₿i	°Po	^₅ At	[®] Rn
Caesium 132.90545196	Barium 137.327		Hafnium 178.49	Tantalum 180.94788	Tungsten 183.84	Rhenium 186.207	Osmium 190.23	Iridium 192.217	Platinum 195.084	Gold 196.966569	Mercury 200.592	Thallium 204.38	Lead 207.2	Bismuth 208.98040	Polonium (209)	Astatine (210)	Radon (222)
⁸⁷ Fr	®Ra	89 - 103 Actinoids	ื₿f	Ďb	[™] Sg	Bh	[™] Hs	Mt	¹¹⁰ Ds	Rg	¹¹² Cn	[™] Nh	[™] FI	™Мс	Lv	¹¹⁷ Ts	[™] Og
Francium (223)	Radium (226)		Rutherfordium (267)	Dubnium (268)	Seaborgium (209)	Bohrium (270)	Hassium (269)	Meitnerium (278)	Darmstadtium (281)	Roentgenium (282)	Copernicium (285)	Nihonium (286)	Flerovium (289)	Moscovium (289)	Livermorium (293)	Tennessine (294)	Oganesson (294)



Structure Field Map Proposed for the Target Reaction:



CO₂ to Methanol Cost Performance Tradeoff

Catalyst Cost



- Synergy between theory and Costs hard to reduce further \bullet In₂O₃+dopants experiment proven Patent space is crowded \bullet Agile ways of combining high throughput simulation and high throughput experiments pay off ln_2O_3 Benefits of AI and Big Data could be harvested Large IP space available \checkmark Low-cost materials developed \checkmark
 - Catalyst Performance

Digitalization in Catalysis and Reaction Engineering Outline



- Introduction –long history of digitalization in science and engineering (OD)
- Automated workflows on the lab scale models for heterogeneous catalysis (OD)
- Feedback loops –from lab to pilot plant (SAS)
- Application of digital tools for scale-up on industrial scale (OD)
- Prediction of materials properties on atom scale using AI for catalyst discovery (SAS)
- Democratization of tools and services: RDM @ NFDI4Cat (SAS)

NFDI₄Cat Vision - Digital Catalysis



A connec Faster inr

A connected community (academia and industry) in catalysis related sciences enabling interlinked knowledge through digital catalysis. Faster innovation through the potential application of artificial intelligence.

A community with an established research data management: research data planning, collection, archiving are carried out according to the FAIR principles by providing repositories, tools, services, and training.

Easy exchange of homogeneously annotated data and structured knowledge through the creation of shared vocabularies, the use of ontologies and persistent identifiers.

Ensuring the lifelong availability of catalysis research data by establishing a shared and understandable knowledge base.



The Value Chain of Digital Catalysis

We are committed to realise the paradigm shift towards digital catalysis in catalysis related sciences along the data value chain and aligned with the real value chain "from molecule to chemical process".

12.06.2023



Open Science and Digitalisation

the drivers in *catalysis* and *chemical engineering*







NFDI₄Cat – the digital enabler for sustainable production of chemicals and energy carriers



Olaf Deutschmann and Stephan A. Schunk Digitalization in Catalysis and Reaction Engineering: More than just a Buzzword?

NFDI₄Cat – Enabler for Enhanced Catalysis Development



NATIONAL Sciences ACADEMIES Medicine

The Importance of Chemical Research to the U.S. Economy



The urgent need to tackle sustainability-related challenges is driving the **need to** improve the pace and efficiency of discovering new catalysts and developing new catalytic processes. Toward this end, available materials, adsorption, and **reaction data** are being harnessed using high-throughput computation and ML to gather further insights and predict new materials (Bo et al., 2018). However, this knowledge extraction approach is challenged by a data deficit, or non-uniform reporting of materials-related data that are not computer readable (Himanen et al., 2019). Wulf et al. (2019) note that "in order to make data widely useful, rather advanced and wellcoordinated approaches are needed that are beyond what a single group or institution can develop and sustain." Digitalization of the catalysis field is essential to "enable efficient data-driven interdisciplinary development of catalysts and catalytic processes" (Wulf et al., 2019). The NFDI4Cat (National Research Data Infrastructure for Catalysis-Related Sciences), supported by the German government, is an example of such a large-scale effort. Establishment of an "internet of catalysis," will guide research along the development chain from molecules to chemical processes.

The creation of digital workflows that bridge theory and experimental studies in catalyst design, characterization, kinetics, and related engineering aspects will accelerate discovery and innovation in the catalysis sciences.

Digitalization in Catalysis and Reaction Engineering: More than just a Buzzword!

Acknowledgements



Thank you for Thank attention







Collaboration and financial support



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